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## An Expeditious Synthesis of Ostopanic Acid, a Plant Anticancer Agent

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**Abstract :** A new and stereoselective synthesis is described as an easy route to ostopanic acid using a versatile reagent : (2E,4E)-5-bromopentadienal. © 1999 Elsevier Science Ltd. All rights reserved.

*Keywords:* Antitumour compounds; polyenes; polyenals; polyenones.

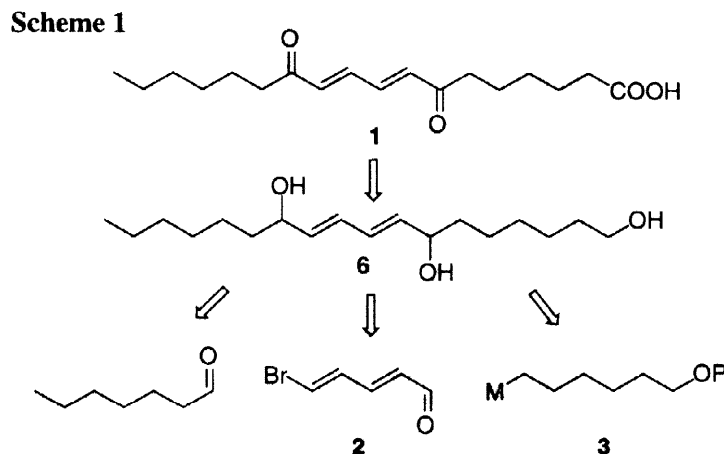
The conjugated polyene structure is found in many natural products<sup>1,2</sup> and it has been shown that the double bond geometry is essential to their biological activity. Fatty acids possessing such a conjugated polyene structure occur in plants. One of them, ostopanic acid **1**<sup>3</sup> isolated from stems and fruits of *Ostodes Paniculata Blume* (Euphorbiaceae) inhibits the growth of P-388 lymphocytic leukemia test system in vitro.

To our knowledge, three total syntheses<sup>4-6</sup> have been reported so far in the literature. They feature from *n*-hexyl furane (12% yield),<sup>4</sup> from (1E,3E)-1,4-bis(trimethylsilyl)-1,3-butadiene (53% yield)<sup>5</sup> or from 4-pentynal (47% yield).<sup>6</sup> In addition, two syntheses of ethyl ostopanate from furfural (16% yield)<sup>7</sup> and from diethyl pimelate (57% yield)<sup>8</sup> have also been described.

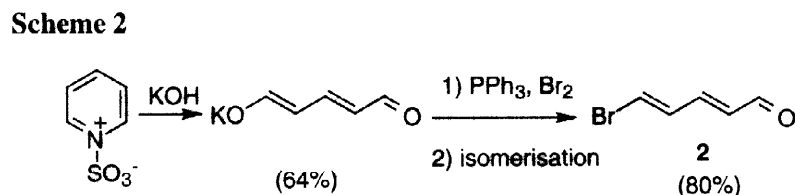
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The E,E-dienyl diketone skeleton structure of ostopanic acid **1**, as well as its biological activity, prompted us to develop a synthetic pathway relying on (2E,4E)-5-bromopentadienal **2**<sup>9-11</sup> to introduce the central polyenic pattern with the right configuration in one step (scheme 1).

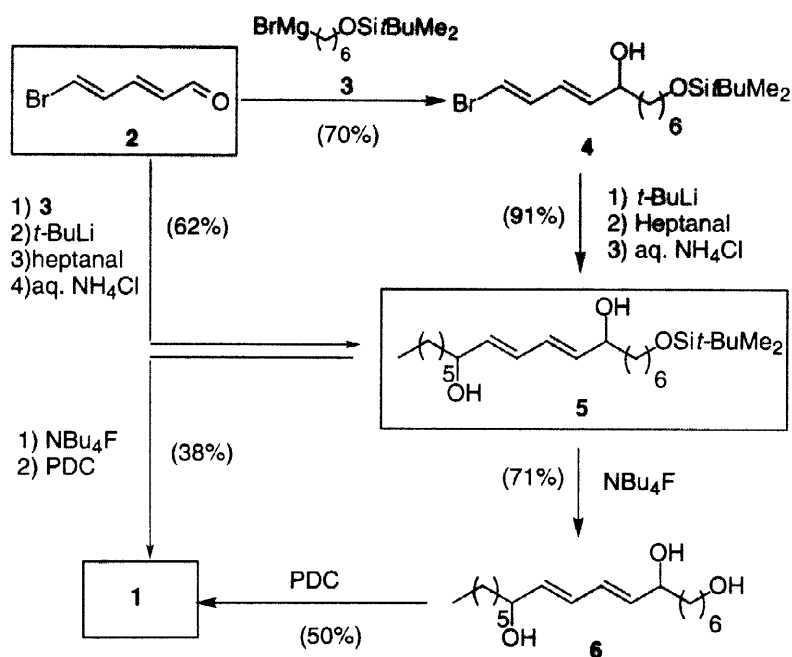


All trans 5-bromopentadienal **2** is obtained by bromination of the potassium salt of glutaconaldehyde with the bromine-triphenylphosphine complex (scheme 2). The two stereoisomers formed (2E,4E/2E,4Z : 75/25) can be easily separated by fractional crystallization (**2** crystallizes in Et<sub>2</sub>O at 0°C), or by flash chromatography.<sup>9,10</sup> Moreover, we have recently improved the access to all trans **2** by isomerizing its (2E,4Z) isomer in quantitative yield.<sup>11</sup>



The first step of the synthesis is the condensation of the organometallic compound **3**<sup>12</sup> on the aldehyde **2** to give the bromohydroxydiene **4** with an all trans configuration. Bromine lithium exchange reaction, followed by condensation with heptaldehyde led to the eighteen carbon atom skeleton of ostopanic acid (*viz.* the monoprotected triol **5**). Deprotection of the primary alcohol of **5** was carried out using tetrabutylammonium fluoride in THF. Then pyridinium dichromate (PDC) oxidation of the resulting trienol **6** led directly to ostopanic acid **1** (scheme 3).

Scheme 3



To improve this procedure we developed a one pot synthesis of **5** from **2**. Furthermore after deprotection of **5**, PDC oxidation can be performed with the crude product **6** without purification. Accordingly, monoprotected triol **5** was obtained in 62% yield, similar to the overall yield of the two step procedure. The transformation of **5** into ostopanone **1** performed in the same pot occurred in 38% yield against 36% for the two-step procedure (scheme 3).

In short, we have developed an expeditious two-step stereocontrolled synthesis of ostopanone **1** from (2E,4E)-5-bromopentadienal **2** with an overall 23% yield.

## EXPERIMENTAL SECTION

230-400 Mesh silica gel was used for flash column chromatography.  $^1\text{H}$  NMR data were recorded at 200 MHz or 400 MHz and  $^{13}\text{C}$  NMR data at 50 MHz or 100 MHz. Mass spectra were recorded by electronic impact at 70 eV. Transmission IR spectra were recorded on FT-IR instrument.

### 6-Bromo-1-tert-butyl dimethylsilyloxyhexane

Under argon, TBDMSCl (3.25 g, 21.55 mmol) was added to 6-bromohexan-1-ol (3.00 g, 16.58 mmol) and imidazole (2.60 g, 38.13 mmol) in dry DMF (100 ml). After stirring at room temperature for 2 h the reaction mixture was washed with saturated  $\text{Na}_2\text{CO}_3$  (50 ml), extracted with light petroleum ether and dried over  $\text{MgSO}_4$ . After removal of the solvent, the residue was chromatographed (eluted with petroleum ether/ $\text{Et}_2\text{O}$  : 90/10) to give a colorless oil (4.63 g, 15.69 mmol, 96 %). GC and MS analyses indicated a partial bromine chlorine exchange:

6-bromo-1-*tert*-butyldimethylsilyloxyhexane / 6-chloro-1-*tert*-butylsilyloxyhexane : 88/12.  $^1\text{H}$  NMR ( $\text{C}_6\text{D}_6$ ) (200 MHz)  $\delta$  : 3.46 (t, 2H,  $J=6.0$ ), 3.09 (t, 0.24H,  $J=6.7$ ,  $\text{CH}_2\text{Cl}$ ), 2.93 (t, 1.76H,  $J=6.9$ ,  $\text{CH}_2\text{Br}$ ), 1.47 (m, 4H), 1.15 (m, 4H), 0.97 (s, 9H), 0.05 (s, 6H) ppm,  $^{13}\text{C}$  NMR ( $\text{C}_6\text{D}_6$ ) (50 MHz)  $\delta$  : 62.96, 33.52, 32.94, 32.85, 28.06, 26.12, 25.24, 18.45 ppm ; IR ( $\text{cm}^{-1}$ ), 2932, 2858, 1472, 1462, 1256, 1104.

#### 1-Bromo-5-hydroxy-11-*tert*-butyldimethylsilyloxyundeca-1,3-diene (4)

A solution of 6-bromo-1-*tert*-butyldimethylsilyloxyhexane (1.03 g, 3.49 mmol) in dry THF (3.4 mL) was added slowly to Mg (0.13 g, 5.35 mmol). After stirring at 45°C for about 3 h, the solution was titrated ( $\text{I}_2 / \text{Na}_2\text{S}_2\text{O}_3$ ,  $\text{C}=0.65$  M). Then, a solution of all trans 5-bromopentadienal **2** (0.35 g, 2.17 mmol) in dry THF (1 mL) was added at -10°C and stirred for 2h at 0°C. The mixture was quenched with aqueous 10%  $\text{NH}_4\text{Cl}$  (4 mL), extracted with  $\text{Et}_2\text{O}$  ; the organic layer was dried over  $\text{MgSO}_4$ , filtered and reduced in vacuo. Purification of the residue by flash chromatography (eluted with light petroleum ether /  $\text{Et}_2\text{O}$  : 70/30) gave **4** as a yellow syrup (0.57 g, 70 %).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) (200 MHz)  $\delta$  : 6.52 (dd, 1H,  $J=10.8$ ,  $J=13.4$ ), 5.91 (d, 1H,  $J=13.5$ ), 5.79 (dd, 1H,  $J=10.8$ ,  $J=15.2$ ), 5.34 (dd, 1H,  $J=6.0$ ,  $J=15.2$ ), 3.75 (m, 1H), 3.56 (t, 2H,  $J=6.2$ ), 1.35 (m, 10H + OH), 1.00 (s, 9H), 0.08 (s, 6H) ppm ;  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) (50 MHz)  $\delta$  : 137.52, 136.77, 127.10, 108.58, 71.89, 63.10, 36.97, 32.64, 29.22, 25.89, 25.213, 18.24, 5.34 ppm; IR ( $\text{cm}^{-1}$ ), 3368, 2930, 2856, 1584, 1472, 1462, 1256, 1100 ; MS (EI 70eV)  $m/z$  (rel. int) 378 ( $\text{M}^+ + 2$ , 4), 376 ( $\text{M}^+$ , 4) 321 (16), 303 (59), 279 (9), 239 (33), 205 (17), 173 (45), 145 (100), 119 (81), 93 (91), 65 (90). Anal. Calcd for  $\text{C}_{17}\text{H}_{33}\text{O}_2\text{BrSi}$  : C, 54.10 ; H, 8.81. Found : C, 54.02 ; H, 8.44.

#### 7,12-Dihydroxy-1-*tert*-butyldimethylsilyloxyoctadeca-8,10-diene (5)

##### a) From 4

A solution of freshly titrated  $^{13}\text{C}$  *t*-BuLi (0.7 mL, 1.77 M in pentane, 1.24 mmol) was added under argon to a solution of bromohydroxydiene **4** (0.13 g, 0.34 mmol) in dry  $\text{Et}_2\text{O}$  (4.0 mL) at -76°C and stirred for 1.5 h. Heptaldehyde (0.09 g, 0.79 mmol) in dry  $\text{Et}_2\text{O}$  (1.0 mL) was added at the same temperature, the mixture was stirred at 0°C for 4 h then quenched with aqueous 10%  $\text{NH}_4\text{Cl}$  (3.0 mL) and extracted with  $\text{Et}_2\text{O}$ . The combined organic layers were dried ( $\text{MgSO}_4$ ), filtered and reduced in vacuo. The residue purified by flash silica gel chromatography (eluted with light petroleum ether /  $\text{Et}_2\text{O}$  : 30/70) afforded the dihydroxydiene **5** as a yellow syrup (0.13 g, 91 %).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) (200 MHz)  $\delta$  : 6.18 (m, 2H), 5.69 (m, 2H), 4.11 (m, 2H), 3.57 (t, 2H,  $J=6.2$ ), 1.41 (m, 20H + 2OH), 0.88 (s, 12H), 0.03 (s, 6H) ppm;  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) (50 MHz)  $\delta$  : 136.35, 129.48, 72.27, 63.12, 37.18, 32.67, 31.70, 29.26, 29.16, 25.85, 25.64, 25.30, 22.50, 18.22, 13.97 ppm; IR ( $\text{cm}^{-1}$ ), 3354, 2928, 2856, 1464, MS (EI, 70eV),  $m/z$  (rel. int), 394 (16), 338 (100), 309 (16), 245 (47), 223 (21), 185 (81), 161 (59), 135 (93), 113 (95), 91 (87), 59 (81). Anal. Calcd for  $\text{C}_{24}\text{H}_{48}\text{SiO}_3$  : C, 69.84 ; H, 11.72. Found : C, 69.84 ; H, 11.52.

##### b) One pot procedure from 2

A solution of 6-bromo-1-*tert*-butyldimethylsilyloxyhexane (bromoether / chloroether : 88/12) (1.40 g, 4.74 mmol) in dry THF (1.5 mL) was added to Mg (0.25 g, 10.30 mmol) in dry THF (0.5 mL), the mixture was refluxed for 2 h under argon. The solution was titrated ( $\text{I}_2 / \text{Na}_2\text{S}_2\text{O}_3$ ,  $\text{C}=1.24$  M). A solution of all trans 5-

bromopentadienal **2** (0.40 g, 2.48 mmol) in dry THF (1.5 mL) was added at  $-10^{\circ}\text{C}$  and stirred for 2 h. Dry THF (4.0 mL) and a solution of *t*-BuLi (4.1 mL, 1.7 M in pentane, 7.0 mmol) were successively added at  $-76^{\circ}\text{C}$  stirred for 1.5 h. Heptaldehyde (0.51 g, 4.47 mmol) in dry THF (2.0 mL) was added at the same temperature, the mixture was stirred at  $0^{\circ}\text{C}$  for 1.5 h. After quenching with aqueous 10%  $\text{NH}_4\text{Cl}$  solution (10 mL), the reaction mixture was extracted with  $\text{Et}_2\text{O}$ , dried ( $\text{MgSO}_4$ ), filtered and reduced in vacuo. Flash silica gel chromatography (eluted with petroleum ether /  $\text{Et}_2\text{O}$  : 50 / 50) yielded 0.63 g (62 %) of dihydroxydiene **5**.

#### (8E,10E)-Octadeca-8,10-dien-1,7,12-triol (**6**)

A solution of  $\text{NBu}_4\text{F}$  (0.57 mL, 1M in THF, 0.57 mmol) was added under argon to a solution of **5** (0.21 g, 0.51 mmol) in dry THF (0.5 mL) and stirred at room temperature for 6h. Saturated aqueous  $\text{NaHCO}_3$  (1 mL) was then added. The mixture was extracted with  $\text{Et}_2\text{O}$ , dried ( $\text{MgSO}_4$ ) and evaporated in vacuo. Crystallization at  $0^{\circ}\text{C}$  from  $\text{Et}_2\text{O}$  gave triol **6** (0.11 g, 72 %).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) (200 MHz)  $\delta$  : 6.17 (m, 2H), 5.67 (m, 2H), 4.08 (dd, 2H,  $J=6.3$ ,  $J=7.1$ ), 3.60 (t, 2H,  $J=6.5$ ), 1.10-1.65 (m, 23H), 0.85 (t distorted, 3H) ppm.

#### Ostopanic acid (**1**)

##### a) From **6**

Under argon, PDC (2.20 g, 5.88 mmol) was added to a solution of triol **6** (0.25 g, 0.84 mmol) in DMF (6.5 mL) at  $0^{\circ}\text{C}$ . After stirring for 5 h at  $0^{\circ}\text{C}$  then 15 h at room temperature, water (80 mL) was added, extracted with  $\text{Et}_2\text{O}$ . The solution was dried ( $\text{MgSO}_4$ ) and reduced to give a white solid which was crystallized at  $0^{\circ}\text{C}$  from  $\text{Et}_2\text{O}$  to give ostopanic acid **1** (0.13 g, 50 %).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) (400 MHz)  $\delta$  : 7.19 (m, 1H), 7.13 (m, 1H), 6.49 (m, 1H), 6.45 (m, 1H), 2.60 (t, 2H,  $J=7.5$ ), 2.57 (t, 2H,  $J=7.5$ ), 2.35 (t, 2H,  $J=7.4$ ), 1.63 (m, 6H), 1.26 (m, 8H), 0.87 (t distorted, 3H) ppm; MS (EI, 70eV),  $m/z$  (rel. int), 308 ( $\text{M}^+$ , 50), 290 ( $\text{M}^+ - \text{H}_2\text{O}$ , 10), 223 ( $\text{M}^+ - \text{C}_6\text{H}_{13}$ , 27), 210 (40), 195 (55), 165 (100), 138 ( $\text{C}_9\text{H}_{14}\text{O}^+$ , 88), 123 (95), 95 (85), 81 (80), 55 (60). m.p. :  $133^{\circ}\text{C}$ , (lit.:<sup>4</sup>  $132\text{--}133^{\circ}\text{C}$ ).

##### b) Direct procedure from **5**

A solution of  $\text{NBu}_4\text{F}$  (2.7 mL, 1M in THF, 2.70 mmol) was added under argon to a solution of silyloxy ether **5** (0.57 g, 1.38 mmol) in dry THF (1.5 mL) and stirred at room temperature for 12 h. Saturated aqueous  $\text{NaHCO}_3$  (5 mL) was added. The reaction mixture was extracted with  $\text{Et}_2\text{O}$ , dried ( $\text{MgSO}_4$ ) and reduced in vacuo. The residue was diluted in DMF (7 mL) and PDC (3.45 g) was added at  $0^{\circ}\text{C}$  under argon, stirred for 12 h at this temperature and then for 10 h at room temperature. Water (90 mL) was added. The mixture was extracted with  $\text{Et}_2\text{O}$ , dried ( $\text{MgSO}_4$ ), filtered and reduced in vacuo. Crystallization at  $0^{\circ}\text{C}$  from  $\text{Et}_2\text{O}$  gave ostopanic acid **1** (0.16 g, 38 %).

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